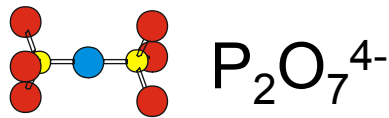


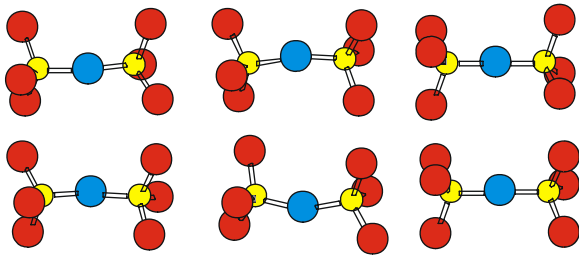
# **Synchrotron X-ray Diffraction and the Solid State Chemistry Community**

Patrick Woodward  
Department of Chemistry  
Ohio State University

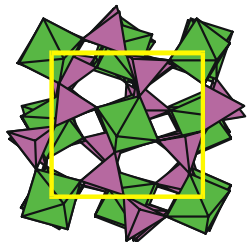
# ZrP<sub>2</sub>O<sub>7</sub> Structural Frustration



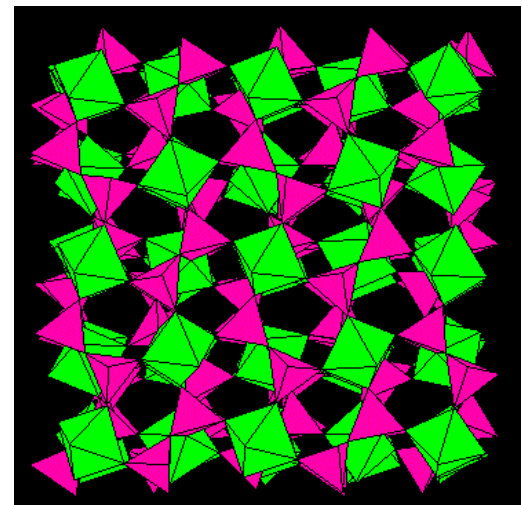
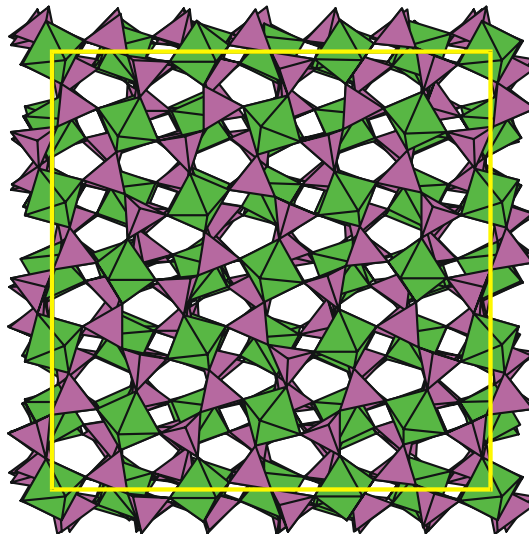
Cool



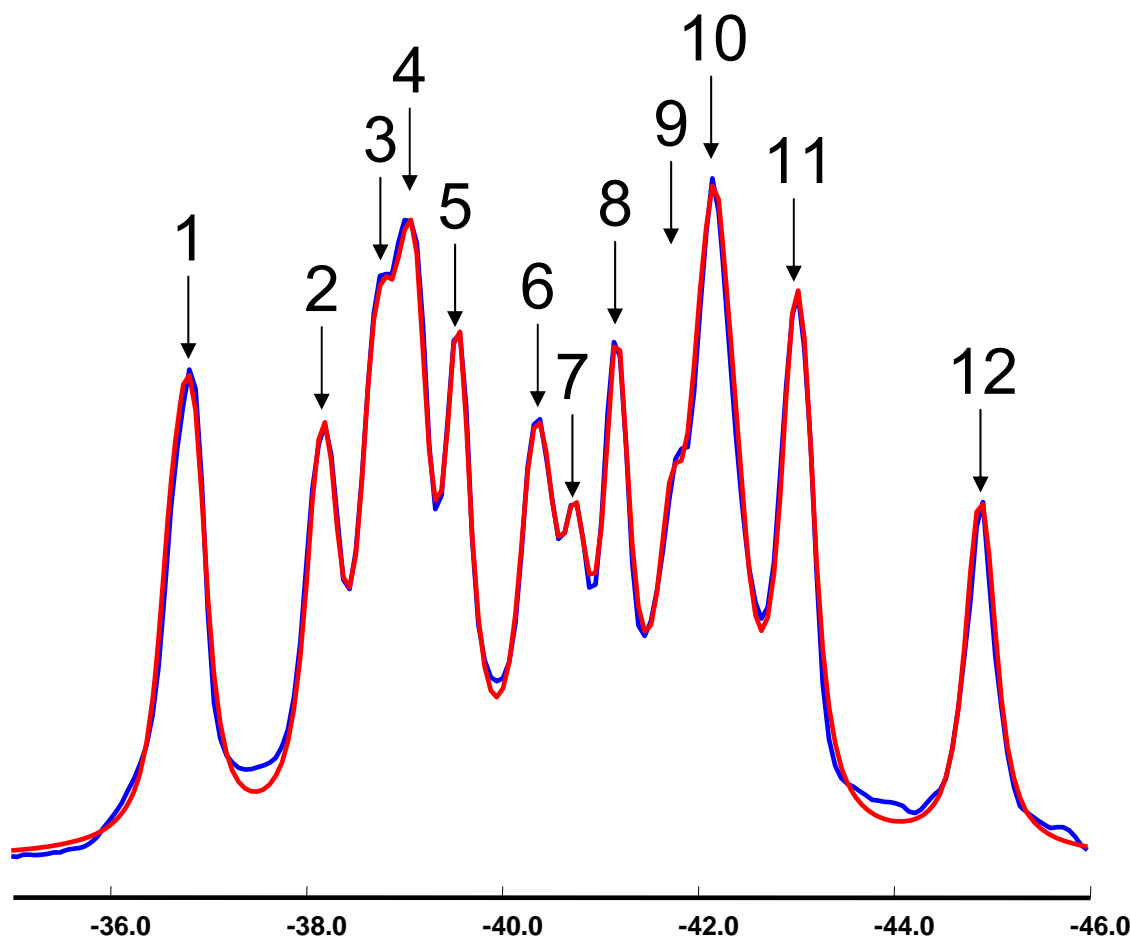
- P-O-P bonds bend
- Symmetry of material is reduced
- Supercell formed
- Pa-3 1x1x1 to Pa-3 3x3x3
- 50 atoms in asymmetric unit
- 11 unique P sites



Cool

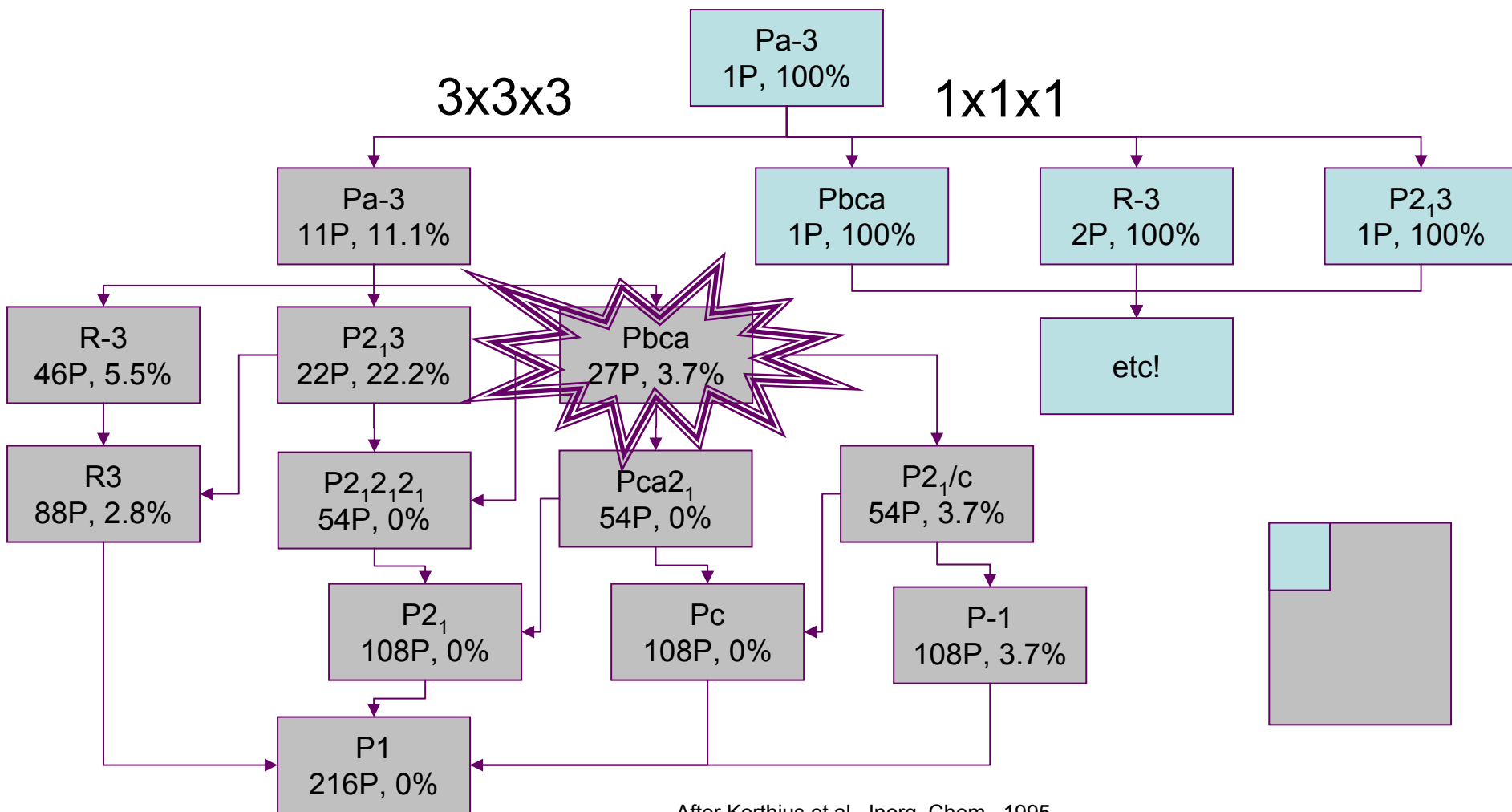


# $\text{ZrP}_2\text{O}_7$ 1D $^{31}\text{P}$ Solid State NMR



- 1D  $^{31}\text{P}$  NMR MAS 10 MHz
- Shows a number of unique phosphorus sites
- At least 12 sites visible
- Expect only 11
- Ian King, Franck Fayon

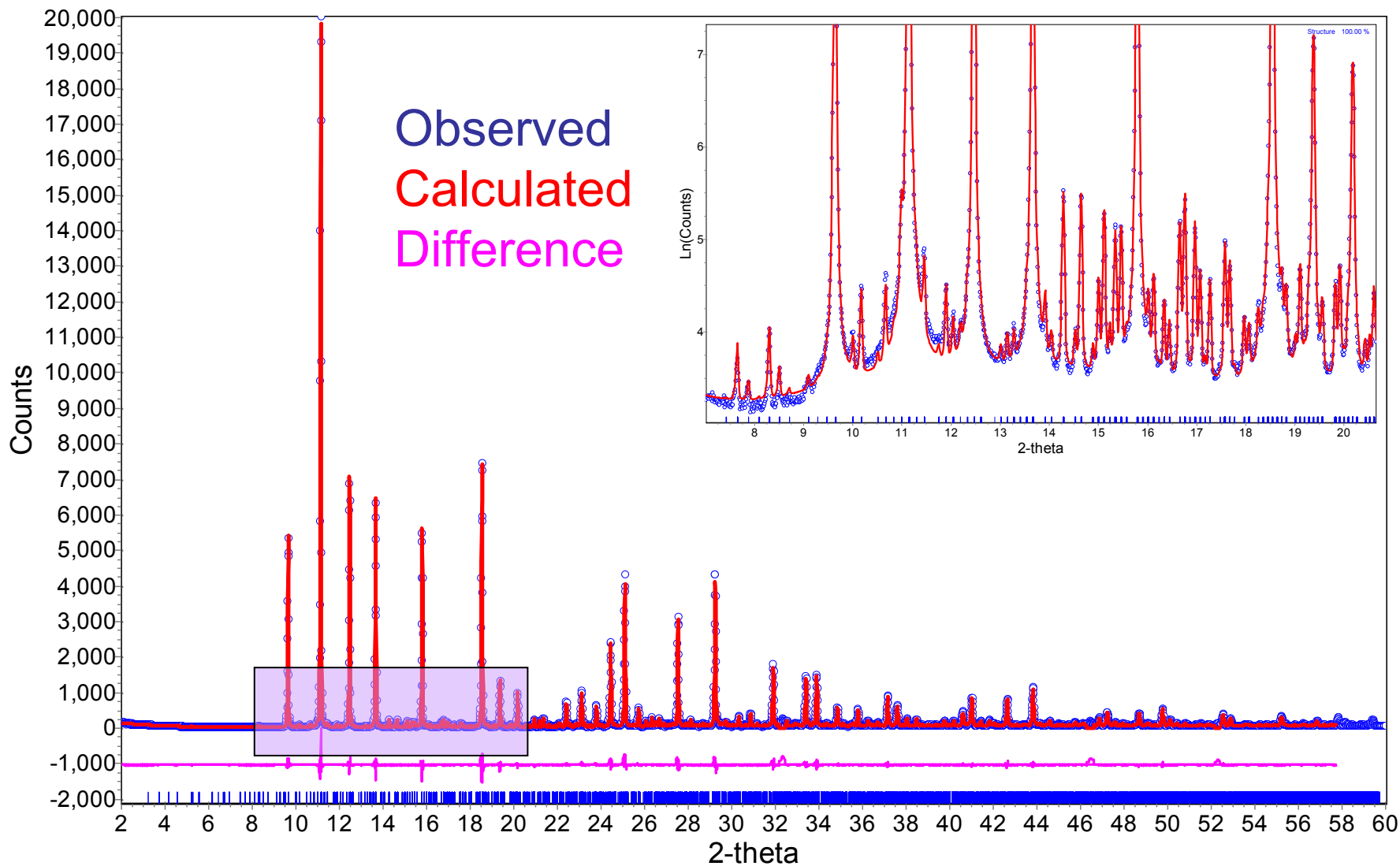
# ZrP<sub>2</sub>O<sub>7</sub> Phase Transitions



After Korthius et al., Inorg. Chem., 1995

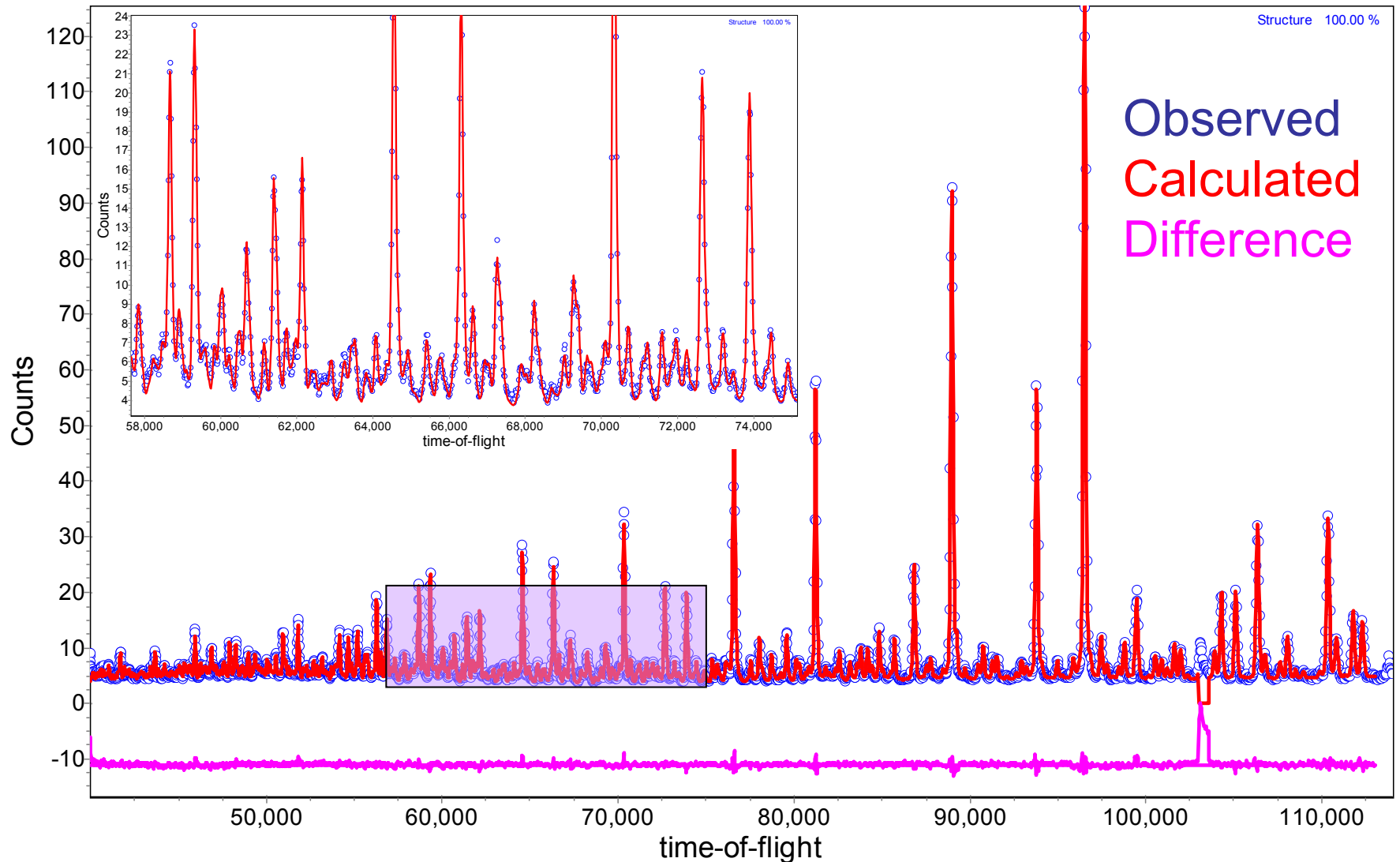
Courtesy of John Evans at Durham University

# ZrP<sub>2</sub>O<sub>7</sub> X ray Rietveld Plot



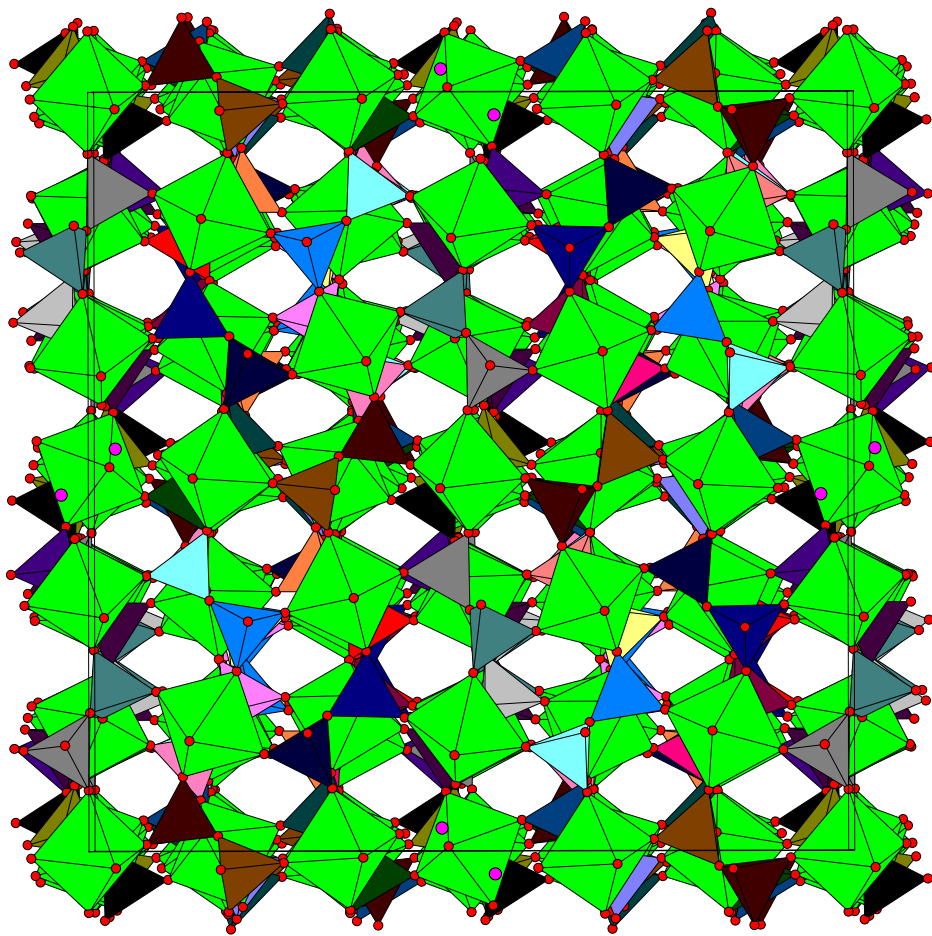
Courtesy of John Evans at Durham University

# ZrP<sub>2</sub>O<sub>7</sub> Neutron Rietveld Plot



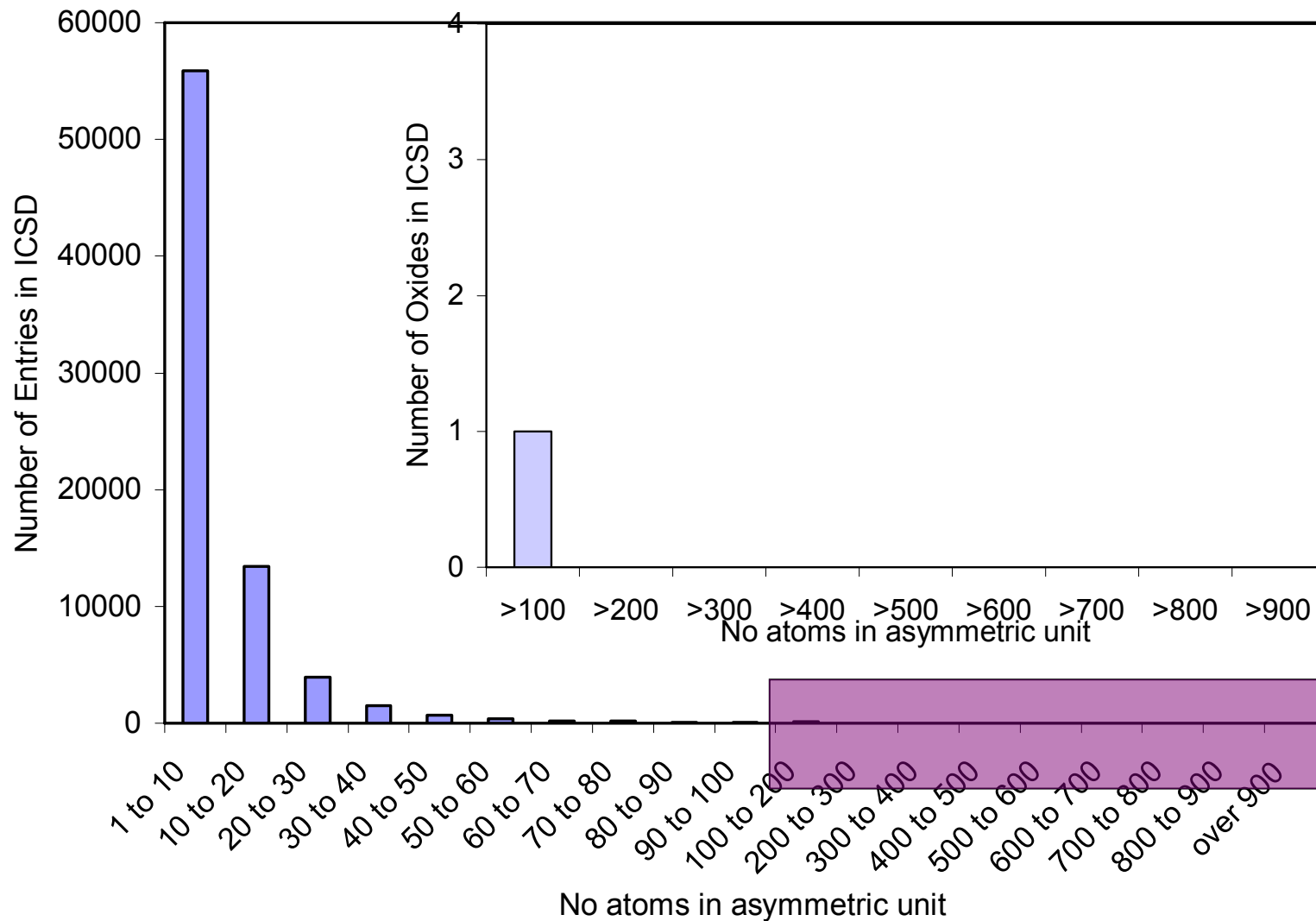
Courtesy of John Evans at Durham University

# ZrP<sub>2</sub>O<sub>7</sub> Structure



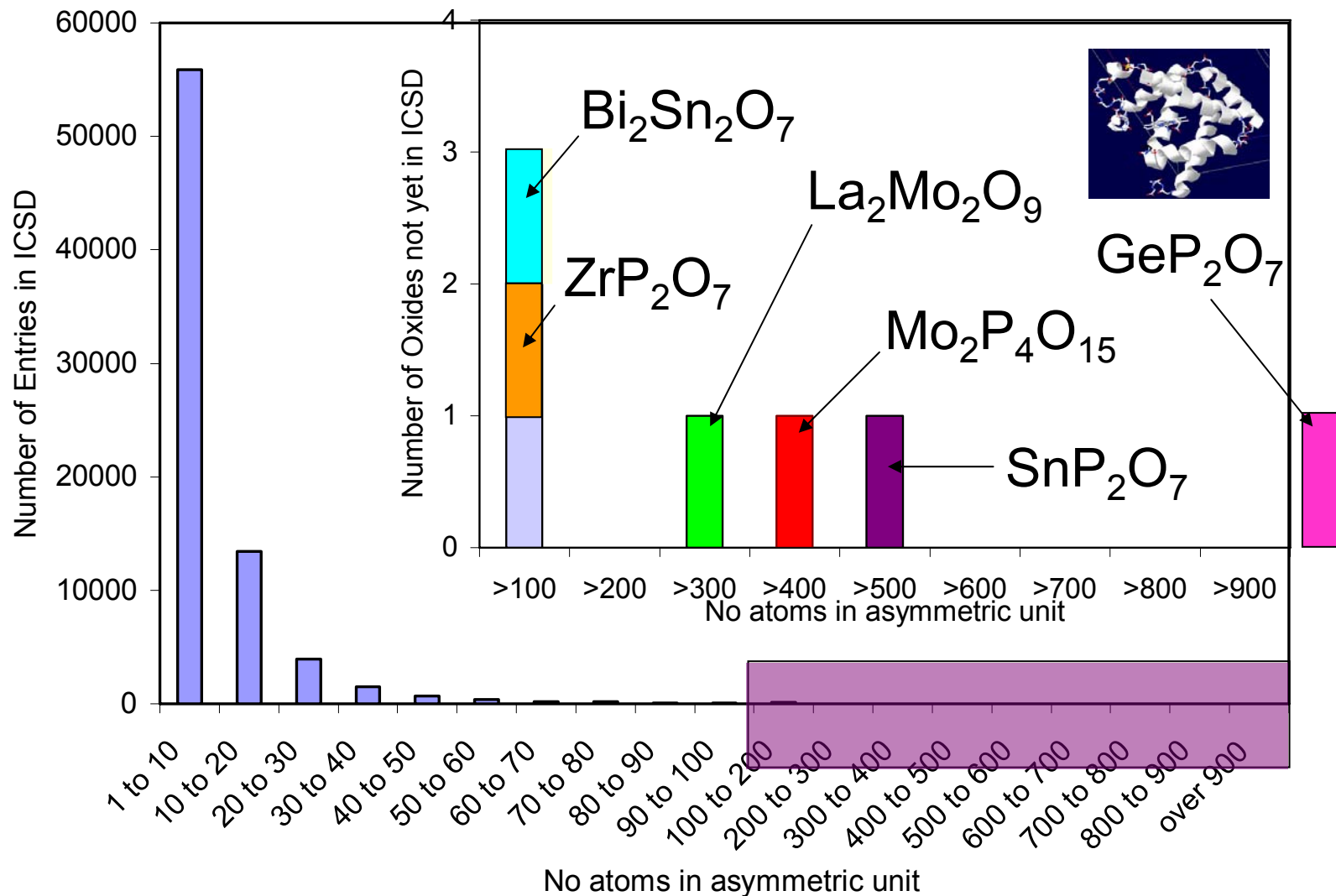
- Pbc<sub>a</sub>
- $a = 24.7437 \text{ \AA}$
- $b = 24.7258 \text{ \AA}$
- $c = 24.7507 \text{ \AA}$
- 136 atoms in asymmetric unit
- 27 P
- 14 Zr
- Errors  $< 0.06 \text{ \AA}$

# Structural Complexity: Oxides





# Structural Complexity: Oxides





## *Complex Octahedral Tilting in a Cryolite Structure*

**T > 1320 K**

**Space group:** Fm3m (Cubic)

**Unit Cell:**

$$a \cong 8.5 \text{ \AA}$$

**Asymmetric Unit:** 4 Atoms

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### Broken Connectivity Tilting

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**470 K < T < 1320 K**

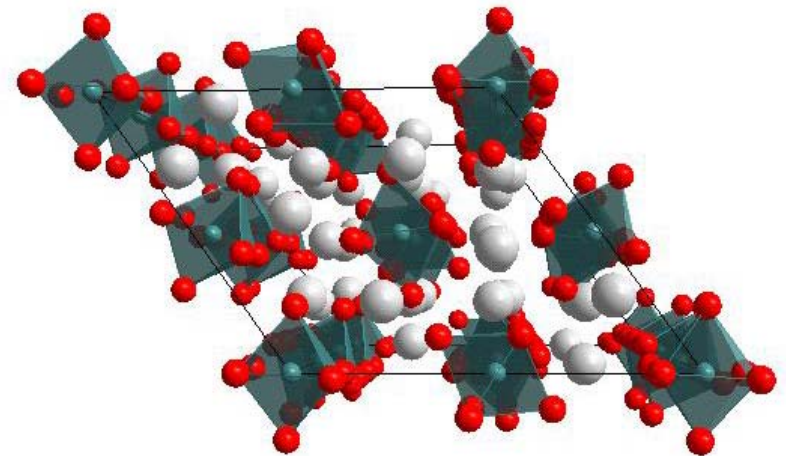
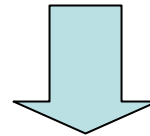
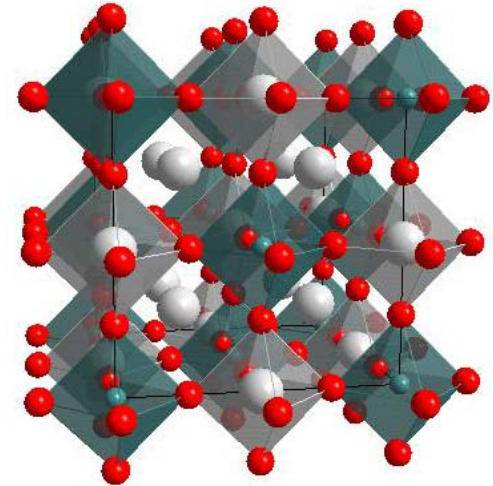
**Space group:** A2/m (Monoclinic)

**Unit Cell:**

$$a = 11.86 \text{ \AA}, b = 17.72 \text{ \AA}, c = 10.13 \text{ \AA},$$

$$\alpha = 90^\circ, \beta = 125.5^\circ, \gamma = 90^\circ$$

**Asymmetric Unit:** 20 Atoms





## *Complex Octahedral Tilting in a Cryolite Structure*

**470 K < T < 1320 K**

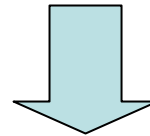
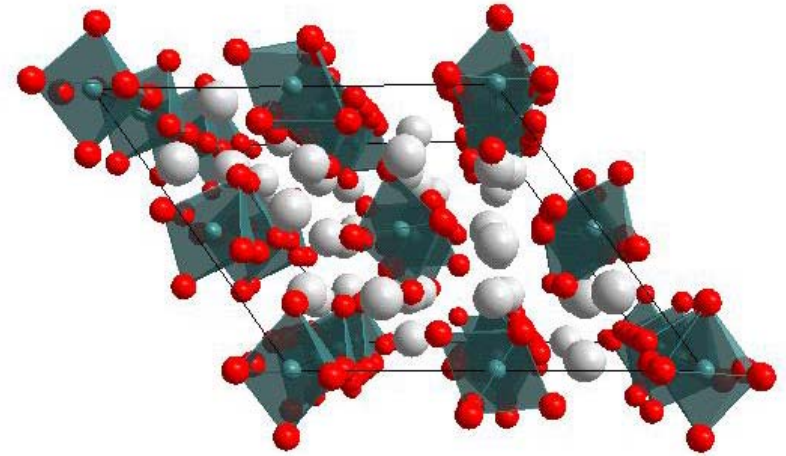
**Space group:** A2/m (Monoclinic)

**Unit Cell:**

$a = 11.86 \text{ \AA}$ ,  $b = 17.72 \text{ \AA}$ ,  $c = 10.13 \text{ \AA}$ ,

$\alpha = 90^\circ$ ,  $\beta = 125.5^\circ$ ,  $\gamma = 90^\circ$

**Asymmetric Unit:** 20 Atoms



## **Broken Connectivity Tilting**

**T > 470 K**

**Space group:** P-1 (Triclinic)

**Unit Cell:**

$a = 11.82 \text{ \AA}$ ,  $b = 17.65 \text{ \AA}$ ,  $c = 10.16 \text{ \AA}$ ,

$\alpha = 89.79^\circ$   $\beta = 125.95^\circ$   $\gamma = 90.41^\circ$

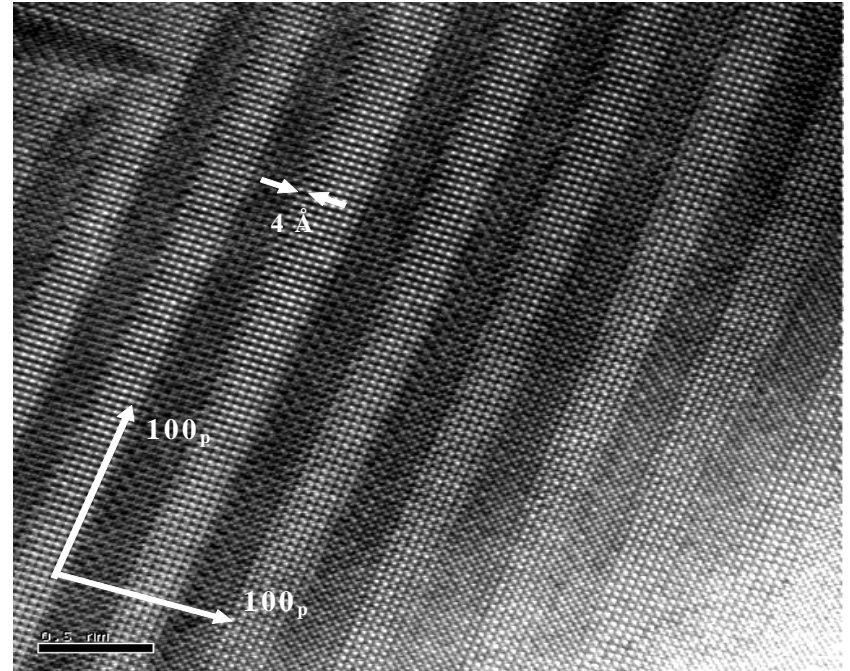
**Asymmetric Unit:** 64 Atoms



# NaLaMgWO<sub>6</sub> Electron Microscopy



Electron Diffraction Pattern

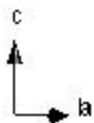
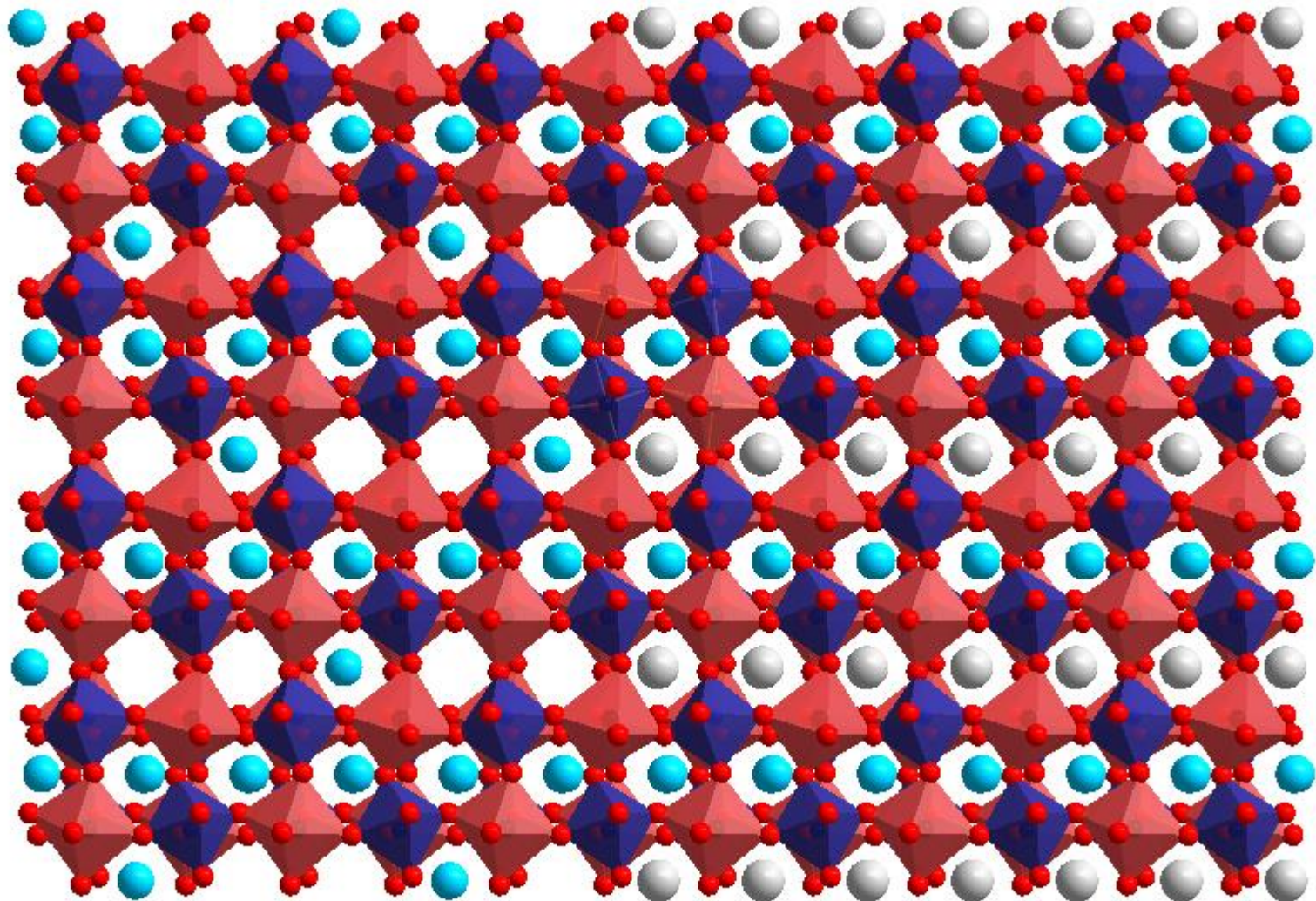


HRTEM Image

Suggests a  $12a_p$  unit cell repeat in the  $a$ -direction  
(perpendicular to the Na/La ordering)



# Proposed Model – Periodic Phase Separation in $\text{Na}_{1-x}\text{La}_{1+x/3}\text{MgWO}_6$



# Comments from Solid State Community

- **Hanno zur Loye (University of South Carolina)**
  - *It would be nicest if we could simply send a capillary containing our powder and have someone run it and send us the data back. Variable temperature capabilities, to look for phase transitions would be great (though I realize this cannot be done via mail in). Having some decent computers there for data collection, data conversion and analysis would be helpful. Maybe we can graduate from VMS.*
- **Svilen Bobev (University of Delaware)**
  - *Advanced capabilities at NSLS to do high resolution powder and single-crystal diffraction. Special environments for low/high  $T$  and under magnetic field will be fantastic.*
- **Mike Lufaso (University of North Florida)**
  - *One aspect that I would find particularly useful is the implementation of a 'mail-order program' for routine powder diffraction measurements.*

# Comments from Solid State Community

- **Ram Seshadri (UCSB)**
  - *Temperature control, particularly below 10 K is what all diffractometers miss that we need.*
- **Cora Lind (University of Toledo)**
  - *A setup with a highly focused beam (for the small opening of the DAC), ability to choose a short wavelength (to get data with a 22 degree cone opening), and most importantly a good detector (we have to run a gazillion data corrections every time before we can do anything useful) are highly desirable.*

# Comments from Solid State Community

- **Ken Poeppelmeier (Northwestern University)**
  - *a purposeful, "user interface" is needed from providing stipend resources, travel funds, to an effective software, web-friendly interface ( much as we do with the journals)....this MUST be built in from the get-go...not left as a promise that gets underfunded or not funded at all down the road. There simply are no funds in the grants of the PI to do this !!!*
- **Ray Schaak (Penn State University)**
  - *One of my main interests in this area would be analyzing structures of nanoscale solids, most likely PDF-type analysis. We've not done it yet, but that is a very definite emerging need. We and others sometimes observe the formation of phases, as nanocrystals made at low-ish temperatures, that are "new" or typically only observed under higher temperature or pressure conditions. Probing these structures... will be critical.*



# Take Home Messages

- Many of the challenging problems in solid state chemistry involve superstructures. Keys to solving these problems include:
  - High resolution
  - Low background/excellent signal-to-noise
  - Variable temperature studies are important
- Links with other techniques are critical
  - Neutron diffraction
  - Electron diffraction
  - Solid state NMR
  - Symmetry analysis
- Solid State Chemists Needs
  - Rapid access/mail in for most work
  - Variable temperature/pressure capabilities
  - PDF studies are likely to grow in importance